



# Zinc Assay Kit

Catalog Number KA1619

250 assays

Version: 05

Intended for research use only

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## Introduction

### Intended Use

- ✓ Direct Assays: zinc in serum, plasma (no EDTA), urine, saliva etc.
- ✓ Drug Discovery/Pharmacology: effects of drugs on zinc metabolism.
- ✓ Environment: zinc determination in waste water, soil etc.

### Key Features

- ✓ Sensitive and accurate: Uses 50  $\mu\text{L}$  samples. Linear detection range 0.12  $\mu\text{M}$  (0.78  $\mu\text{g/dL}$ ) to 10  $\mu\text{M}$  (65  $\mu\text{g/dL}$ ) zinc in 96-well assay format.
- ✓ Simple and high-throughput: The procedure involves addition of a single working reagent and incubation for 30 min. Can be readily automated as a high-throughput assay for thousands of samples per day.
- ✓ Improved reagent stability and versatility: The optimized formulation has greatly enhanced reagent and signal stability. Cuvette or 96-well plate assay formats possible.
- ✓ Low interference in biological samples. No pretreatments are needed.

### Background

Zinc is an essential trace element and plays many key roles in metabolism. It is required for the activity of more than 300 enzymes, the structure of many proteins, and control of genetic expression. Zinc status affects basic processes of cell division, growth, differentiation, development, performance and aging through its requirement for synthesis and repair of DNA, RNA and protein. The common causes of zinc deficiency are low dietary intakes and low bioavailability. Clinical signs of zinc deficiency include acrodermatitis, low immunity, diarrhea, poor healing, stunting, hypogonadism, fetal growth failure, teratology and abortion. Zinc deficiency has now been recognized to be associated with many diseases such as malabsorption syndrome, chronic liver disease, chronic renal disease, sickle cell disease, diabetes, malignancy, and other chronic illnesses.

Simple, direct and automation-ready procedures for measuring zinc concentration in biological samples are highly desirable in Research and Drug Discovery.

### Principle of the Assay

The Zinc Assay Kit is designed to measure zinc directly in biological samples without any pretreatment. The present method utilizes a chromogen that forms a colored complex specifically with zinc. The intensity of the color, measured at 425 nm, is directly proportional to the zinc concentration in the sample.

## General Information

### Materials Supplied

List of component

Component	Amount
Reagent A	50 mL
Reagent B	1 mL
Reagent C	1 mL
EDTA (100 mM)	1 mL
Zinc standard (50 $\mu$ M)	1 mL

### Storage Instruction

Store Reagents B and C at -20°C and other components at 4°C. Shelf life: 6 months after receipt..

### Materials Required but Not Supplied

- ✓ Pipeting devices and accessories.
- ✓ Clear bottom 96-well plates (e.g. Corning Costar) and plate reader.
- ✓ Cuvettes and Spectrophotometer for measuring OD 425 nm.

### Precautions for Use

- ✓ Reagents are for research use only. Normal precautions for laboratory reagents should be exercised while using the reagents.

## Assay Protocol

### General Considerations

Because the shift in the peak wavelength (from 413 nm to 425 nm) is very small, the color change is not visually evident. Physiological concentrations of other metal ions do not interfere. Zn<sup>2+</sup> chelators (e.g. EDTA, EGTA) should be avoided during sample preparation.

### Reagent Preparation

Equilibrate all reagents to room temperature. Vortex Reagents B and C before assay. Prepare enough Working Reagent: for each assay well, mix 200 µL Reagent A, 4 µL Reagent B and 4 µL reagent C.

### Sample Preparation

Serum and plasma samples should be clear and free of turbidity or precipitates. If present, precipitates should be removed by filtration or centrifugation on a table centrifuge. Prior to assay, dilute serum or plasma samples 5-fold (n = 5) in deionized water.

### Assay Procedure

✓ Procedure using 96-well plate:

1. Prepare standards in deionized water. Transfer 50 µL of the Zn<sup>2+</sup> standards into wells of a clear flat-bottom 96-well plate.

No	STD + H <sub>2</sub> O	Vol (µL)	Zn <sup>2+</sup> (µM)
1	20 µL + 80 µL	100	10.0
2	15 µL + 85 µL	100	7.5
3	10 µL + 90 µL	100	5.0
4	5 µL + 95 µL	100	2.5
5	0 µL + 100 µL	100	0

2. Transfer 50 µL Sample and Sample Blank (50 µL sample + 2 µL EDTA) into wells of a clear flat-bottom 96-well plate. Add 200 µL working reagent and tap plate lightly to mix.
3. Incubate 30 min at room temperature and read optical density at 425 nm (range 420 - 426 nm).

✓ Procedure using cuvette:

1. Transfer 200 µL standards, sample and sample blank (200 µL Sample + 8 µL EDTA) to appropriately labeled tubes.
2. Add 800 µL working reagent and tap lightly to mix.
3. Incubate 30 min and read optical density at 425 nm.

## Data Analysis

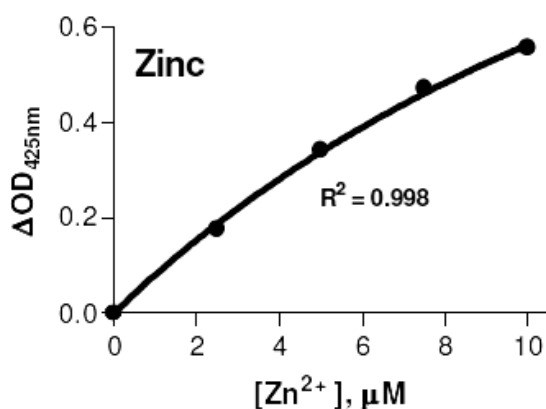
### Calculation of Results

Subtract blank OD (water, #5) from the standard OD values and plot the  $\Delta OD$  against  $Zn^{2+}$  standard concentrations. Calculate  $\Delta OD$  for the Sample (=  $OD_{\text{SAMPLE}} - OD_{\text{SAMPLE BLANK}}$ ). Determine the Sample  $Zn^{2+}$  concentration from the standard curve by non-linear regression fitting with a single-site saturation binding function ( $\Delta OD = a \times [Zn^{2+}] / (b + [Zn^{2+}])$ ). If the  $Zn^{2+}$  concentration is higher than 10  $\mu M$ , dilute sample in deionized water. Repeat the assay and multiply the results by the dilution factor.

Conversions: 1  $\mu M$  zinc equals 6.5  $\mu g/dL$  or 0.065 ppm (65 ppt).

#### ✓ Examples

Samples were assayed in duplicate using the 96-well protocol. The zinc concentration ( $\mu g/dL$ ) was  $45.2 \pm 4.4$  for a human serum,  $49.3 \pm 3.2$  for rat serum,  $35.8 \pm 1.4$  for rat plasma,  $113 \pm 2$  for Invitrogen fetal bovine serum,  $30.8 \pm 1.5$  for human urine,  $0.99 \pm 0.31$  for a sea water sample,  $<0.8$  for a soil extract and a human saliva sample.



Standard Curve in 96-well plate assay